

C. U. SHAH UNIVERSITY

Winter Examination-2022

Subject Name : Qualitative Optical Spectroscopic Method - I

Subject Code : 5SC03QSC1

Branch: M.Sc. (Chemistry)

Semester: 3

Date: 22/11/2022

Time: 11:00 To 02:00

Marks: 70

Instructions:

- (1) Use of Programmable calculator and any other electronic instrument is prohibited.
- (2) Instructions written on main answer book are strictly to be obeyed.
- (3) Draw neat diagrams and figures (if necessary) at right places.
- (4) Assume suitable data if needed.

SECTION – I

- Q-1 Attempt the Following questions (07)**
- a. Why we can't distinguish enantiomers by IR spectroscopy? **01**
 - b. Write the wavenumber range for free and H-bonded O-H group. **01**
 - c. Give the equation showing relation between wavenumber and force constant. **01**
 - d. Define spectroscopy. **01**
 - e. Give wavelength range for Far IR region. **01**
 - f. Give the examples of molecules which active in Raman but not in IR spectroscopy. **01**
 - g. Define Raman scattering. **01**
- Q-2 Attempt all questions (14)**
- A** Explain the sampling techniques used in IR spectroscopy. **07**
- B** Discuss the mechanism of Raman effect by quantum theory. **07**
- OR**
- Q-2 Attempt all questions (14)**
- A** Write a note on various factors affecting vibrational frequency. **07**
- B** Explain the mechanism of Raman scattering by classical theory. **07**
- Q-3 Attempt all questions (14)**
- A** Explain fundamental vibrations for IR spectroscopy. **05**
- B** Discuss the disadvantages of Raman spectroscopy over Infrared spectroscopy. **05**
- C** Give the applications of IR spectroscopy. **04**
- OR**
- Q-3 A** Explain the instrumentation of Dispersive IR spectroscopy. **05**
- B** Discuss the advantages of Raman spectroscopy over Infrared spectroscopy. **05**
- C** Explain resonance Raman technique. **04**



SECTION – II

- Q-4 Attempt the Following questions (07)**
- a. How one can increase population difference as per Boltzman Distribution Law equation? **01**
 - b. Define X-Ray Diffraction. **01**
 - c. Write equation of Larmor frequency for NMR spectroscopy. **01**
 - d. What do you mean by downfield field shift? **01**
 - e. Write any two examples of nuclei having integral spin value. **01**
 - f. Give the Bragg's equation and name of different terms involve in the equation. **01**
 - g. Give example of molecules in which shielding of proton take pace. **01**
- Q-5 Attempt all questions (14)**
- A** Discuss the various factors affecting on chemical shift in NMR. **07**
 - B** Write a note on powder diffraction method in detail. **07**
- OR**
- Q-5**
- A** Explain types detectors used in X-ray diffraction. **07**
 - B** Discuss the continuous wave NMR instrumentation. **07**
- Q-6 Attempt all questions (14)**
- A** Write a note on different monochromators used in X-ray spectrometer. **05**
 - B** Explain spin-spin coupling mechanism for NMR in detail. **05**
 - C** Write the applications of X-rays diffraction. **04**
- OR**
- Q-6 Attempt all Questions**
- A** Discuss the single crystal X-ray diffraction. **05**
 - B** Write a note on chemical shift for NMR spectroscopy. **05**
 - C** Draw labeled instrumental diagram of FT-NMR instrument. **04**

